



## Exploiting Multicomponent Reagents in Synthesis: Stepwise vs Synchronised Cooperative Behaviours

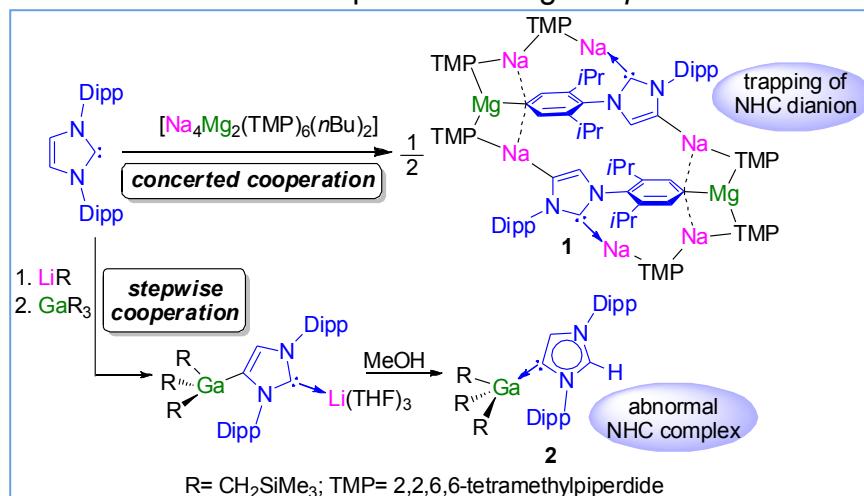
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Multicomponent bimetallic organometallic reagents are capable of producing new chemistry irreproducible by either of their single organometallic components.<sup>[1]</sup> However, recent work<sup>[2]</sup> has revealed that such heterobimetallic reagents do not necessarily need to form mixed-metal compounds in order to exhibit this unique cooperative behaviour, as they can alternatively operate in a stepwise process.

This talk firstly reviews our new results using cooperative bimetallics to promote deprotonative metallations, operating either in a synchronised or stepwise manner. These approaches have established novel ways to trap dianionic NHC (N-heterocyclic carbene) fragments in reactions of unprecedented chemoselectivity (**1** in Figure)<sup>3a</sup> and to access elusive main group abnormal NHC complexes<sup>3b</sup> (**2** in Figure). Secondly, the opening applications of mixed ammonium-magnesiate and lithiate salts in Green Chemistry will be revealed through addition reactions of Grignard reagents to ketones under air and at room temperature using *Deep Eutectic Solvents*.<sup>4</sup>



### References:

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